

Original Article

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The antimicrobial effect of doxycycline and doped ZnO in TiO₂ nanotubes synthesized on the surface of orthodontic mini-implants

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ABSTRACT

Objectives: Anchorage preservation is crucial in orthodontic treatment success. Mini-implants make a revolution in this domain. The failure of orthodontic mini-implants due to inflammation and infection is one of the reasons for anchorage loss. The purpose of this study was to evaluate the effect of a novel mini-implant surface modification to improve resistance against microbial contamination and surrounding tissue inflammation.

Material and Methods: Twenty-four orthodontic mini-implants (Jeil Medical Corporation, Korea) with 1.6 mm diameter and 8 mm length were randomly divided into three groups: Group 1: Control group, Group 2: Nanotubes were made on the surface with anodisation, and Group 3: Zinc Oxide (ZnO) doped into nanotubes, and then doxycycline is added to them. The anti-bacterial efficacy against *Porphyromonas gingivalis* was evaluated using the disk diffusion method. To analyze data, Kruskal–Wallis, Friedman, and Wilcoxon tests were done. The significance level was set at 0.05.

Results: No zone of the inhibition was formed in Groups 1 and 2. In Group 3, the mean (SD) diameter of the inhibition zone in the first 5-day to sixth 5-day were 38.7(8.2), 25(4.8), 17.8(5.6), 7.63(5.37), 1.5(2.83), and 0 millimeters, respectively.

Conclusion: Nanotubes containing doped ZnO and Doxycycline are capable of preventing bacterial growth around the mini implant surfaces for at least up to 30 days. To manage inflammation of surrounding tissues of mini-implants, nanotubes are not effective alone. Therefore, the presence of diffusible materials in addition to nanotubes on the surface of mini-implants is necessary.

Keywords: Orthodontic mini-implant, Nanotube, Titanium, Doxycycline, Zinc oxide

INTRODUCTION

Anchorage preservation is one of the most challenging orthodontic issues defined as resistance to unwanted tooth movement.^[1] In routine intra-oral methods, anchorage loss is always possible and is a great concern and extra-oral methods have limitations due to patient cooperation.^[2]

The introduction of mini-implants as skeletal anchorage has allowed clinicians to overcome anchorage instability or dependence on patient cooperation.^[2-4] Their advantages are the ease of insertion and removal, reasonable cost, biocompatibility, tolerating of most orthodontic forces, and acceptable success rate.^[5,6] They facilitate complex tooth movements, for example, posterior

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teeth movements (intrusion, distalization, and protraction), intrusion of incisors, and cross-bite correction.^[7-9]

The concern in the application of mini-implants is screw loosening and failure of it causes anchorage loss.^[10] Factors related to the survival of mini-implants include host tissuerelated, mini-implant-related, insertion procedure-related, loading-related factors, and patient oral hygiene.^[11] Microbial infection of surrounding soft and hard tissues and following inflammation make bone loss and screw loosening. The role of bacteria as a prominent factor has been confirmed in peri-implantitis.^[12] Soft-tissue integrity around dental implant protects alveolar bone from the oral environment. This barrier can be impaired by microbial penetration and plaque formation on the implant surface.^[13] The following inflammation makes bone resorption and loosening of the implant, so maintaining hygiene around implants and preventing microbial colonization in this area is essential.^[13,14]

Implant surface modification is an effective method to reduce implant-related infections.^[15] Various techniques such as sandblasting, laser therapy, acid etching, hydroxyapatite coating, and anodization have been tested.^[16-22] Anodizing creates the characteristics of regular nano-topography, like nano-tubes. It improves the biomechanical stability of *in vivo* porous titanium implants compared to other types of surface modification techniques.^[23] Increasing surface [Figure 1] and lower contact angles with water create a hydrophilic surface that supports protein adhesion and improves the initial stage of osteointegration.^[24-27] Studies showed that Titanium Oxide (TiO₂) nano-tubes have antibacterial properties against *Streptococcus mutans*, *Escherichia coli*, and *Staphylococcus*.^[15,28] Karmarker *et al.* showed anodizing improved removal torque of orthodontic mini-implants.^[10]

High concentrations of zinc show antibacterial properties.^[29] The antibacterial mechanism of ZnO is mainly the production of reactive oxygen species that, due to their high reactivity, can destroy the integrity of the cell wall and cause bacterial death.^[30] Petrini *et al.* showed that titanium surface chemically modified with ZnO significantly reduced five streptococcus species.^[31] Pharmaceutical agents such as antibiotics add antibacterial properties to implant surfaces.^[13] Doxycycline improves bone



Figure 1: Increase surface nanotubes.

growth and the treatment of periodontal disease as well as peri-implantitis.^[32] Considering that, the failure of miniimplants due to infection and inflammation caused by it, and considering that orthodontic treatments themselves cause significant changes in the bacterial environment of the mouth, which is associated with more gingivitis.^[33,34]

This study aimed to investigate the surface modification and antimicrobial effect on orthodontic mini-implants.

MATERIAL AND METHODS

Twenty-four orthodontic mini-implants (Jeil Medical Corporation, Korea) with 1.6 mm diameter and 8 mm length were randomly divided into three groups: Group 1: Control group, Group 2: nanotubes were made on the surface with anodization, and Group 3: ZnO doped into nanotubes and then Doxycycline is added to them. The anti-bacterial efficacy against *Porphyromonas gingivalis* was evaluated using the disk diffusion method.

Preparing the surface of samples

Nanotube fabrication

Mini-implants in the second and third groups were cleaned by the ultrasound bath with an acetone alcohol solution (2-propanol) and washed with de-ionized water. An anodic oxidation process was performed in a magnetic stirrer glass reactor, in which a mini-implant as anode and platinum electrode was used as cathode. Only the head and half of the screw are immersed in the solution. Two electrodes with a distance of 3 cm were placed in an electrolyte solution composed of ammonium fluoride (NH₄F). The process was continued at 60 v for 6 h at room temperature. The samples were washed using deionized water after the anodization process and dried with nitrogen gas. Then, the anodized samples were of TiO₂ to a crystalline structure.^[35] The surface morphology of samples was evaluated by scanning electron microscopy and see nanotubes of TiO₂ have been created on it [Figure 2].

Incorporation of ZnO nanoparticles into Ti nanotube

In the third group, mini-implants were anodized, and then the ZnO was doped into nano-tubes to reach the concentration of 0.0150 M, then they were immersed in an aqueous solution containing nitrate on $(Zn (NO_3)_2)$ and hexamethylenetetramine $(CH_2)_6N_4$ with a molar ratio of 1:2.2 mg of citric acid was added to the solution. The samples passed hydrothermal reaction for 2 h at 70°C to form all the ZnO-decorated nano-tubes.^[36]

Drug loading

In the following, for the third group, 100 mg (doxycycline-Hyclate D9891, Sigma-Aldrich) to a solution of 10 wt% bovine gelatin (type B powder reagent, CAS 9000-70-8; Sigma-Aldrich) in de-ionized water and stirred for 1 h at 40°C, to obtain a homogenous drug/gelatin solution. Then each mini-implant was sonicated in gelatin/drug solution for 30 min to penetrate the drug/gelatin into nano-tubes. Each loaded mini-implant was left to dry at room temperature for 24 h.^[29]

All samples were exposed to (UV) radiation for sterilization under a UV lamp with a power of 20 w and a distance of 10 cm from the lamp for 1 h.^[28]

Culture of bacteria and antibiogram test

Antibacterial activity of the samples against the strain of P. gingivalis (ATCC 33277) was performed using the disk diffusion method.^[14] Bacteria were cultured in tubes containing 5 ml of growth control medium under anaerobic conditions at 37°C for 48 h to reach bacteria in the midlogarithmic phase. Then, bacterial suspensions were diluted to achieve the final density of 1.5×10^6 colonies per ml. Diluted suspension of bacteria (1 ml) was placed on the agar medium and allowed to dry for 10 min. Then, mini-implants of all three groups were placed on agar blood plates contaminated with bacteria. Plates containing culture medium were incubated under anaerobic conditions at 37°C. Then, the maximum diameter of the inhibitory zone was measured in mm after 5 days.^[30] To evaluate the persistence of antibacterial properties, the samples in which the inhibitory zone was formed were transferred to the new plate every 5 days until no forming of the inhibitory zone was.

RESULTS

In the first and the second group, the inhibitory zone was not formed. In the third group, the mean and standard deviation of the diameter of inhibition zones is shown in



Figure 2: SEM view of titanium oxide nanotubes (×20,000).

[Table 1 and Figure 3]. Kruskal–Wallis test showed that there was a significant difference between the three groups based on different times (the first 5-day to the sixth 5-day). The calculated *P*-value in 5-day period intervals was: First 5-days to fourth 5-day 0.001 >PV, fifth 5-day 0.124 > PV, and sixth 5-days 0.1> PV.

Friedman test showed that there was a significant difference between the first 5-days and the sixth 5-day (0.001 > PV) [Table 2]. Wilcoxon test showed that there was a significant difference between all-time intervals (P < 0.05) except for the fifth 5-day and sixth 5-day (P = 018).

Table	1:	Inhibitory	zone	diameter	(mm)	in	the	third	group
Titaniı	ım	Nano Tube	& Zin	nc (TNT-Z	n-drug) in	5-da	y peri	ods.

	5-Day Periods					
	1 st	2 nd	3 rd	4^{th}	5^{th}	6^{th}
Min.	30	20	10	0	0	0
Max.	48	30	26	15	7	0
Mean	38.7	25	17.8	7.63	1.5	0
Standard deviation	2.83	4.8	5.6	5.37	2.83	0
P-value	0	0	0	0/001	0/124	1



Figure 3: (a) First 5-day: Sample from Group 3 (upper side), a sample from Group 2 (lower left), and sample from Group 1 (lower right). Samples of Group 3 in the following 5-days: (b) Second 5-day, (c) Third 5-day, (d) Fourth 5-day, (e) Fifth 5-day, and (f) sixth 5 days.

Table 2: Test statistics. ^a	
n	8
Chi-square	39.104
Df	5
Asymp. Sig.	0.000
^a Friedman test	

DISCUSSION

Orthodontic mini-implants can provide reinforce anchorage, improve the success rate of treatments, and facilitate complex orthodontic treatments. Their effect on improving the quality of orthodontic treatments depends on their survival. Inflammation of the surrounding tissues of miniimplant is the most important factor in adjacent bone loss, reduction of mechanical retention, and ultimately failure of it. Therefore, any factor that prevents, reduces, or eliminates this inflammation improves the clinical success of miniimplants.^[31,33]

The primary approach is preserving good oral hygiene of extra-tissue parts of mini-implants. In *in vivo* and *in vitro* studies, mechanical removal of bio-films around dental implants by air powder abrasion and titanium brush has not entirely removed the bio-films and about 10% of them have remained.^[34] Therefore, mechanical methods of plaque control around mini implants are not enough to eliminate all bacterial bio-films. Another way is the use of local antimicrobial materials on the tissues around the mini-implant. This depends on the patient's cooperation and is limited only to extra-tissue parts of mini-implants and intra-tissue parts are not in access.^[14] Chen *et al.* showed the local application of hydro-gel containing ibuprofen and fibroblast growth factor can control inflammation and reduce the risk of peri-implantitis.^[31]

Adding anti-inflammatory and antimicrobial agents to the mini-implant surface is the only way to be effective on both intra- and extra-tissue parts.^[15] Creating nano-tubes on the surface of titanium mini-implants by the anodizing method have some advantages, for example, increasing surface area, the lower contact angle with water, antimicrobial properties on the mini-implant surface, improved protein adhesion, osteoblast reaction, and the initial stage of osteointegration.^[10,24,27] Giordano et al. showed that surface anodization reduces the colonization of bacteria such as Staphylococcus aureus, Staphylococcus epidermidis, S. mutans, and P. gingivalis on the surface of implants.^[37] Cui et al. stated that TiO₂/bio-composite layers have excellent antibacterial activity against S. mutans.^[21] Ercan et al. showed that titanium nano-tubes significantly reduced the formation of S. aureus biofilm on the surface after 2 days.^[34] Titanium nanotubes have a high surface/volume ratio and negative electrical

charge on their walls that make antibacterial properties and the potential to bind with positive proteins surface on osteoblasts.^[35] Karmarker *et al.* showed that the maximum removal torque of mini-implants was higher in the anodized group, in an animal study, which promises better anchorage in orthodontic treatments.^[10]

Nano-tubes reduce the colonization of bacteria on the surface of the implant and facilitate primary osteointegration. This reduces infection potential over time.^[34] This means nanotubes act as anti-microbial agents on the surface. However, in the present study, in the control group and mini-implants with nano-tubes, an inhibitory zone has not formed that indicates just the presence of nano-tubes will not inhibit microbial growth in adjacent tissues and far from the surface. This can be due to not presence of diffusible agents in nanotubes in Group 2. Therefore, it is necessary to add diffusible antimicrobial agents, such as ions and drugs, to make and increase antimicrobial properties farther away from the mini-implant surface.

ZnO has anti-bacterial properties.^[29,35] Its mechanism is to produce reactive oxygen species and destroy the integrity of the cell wall of bacteria.^[30] Moreover, the released Zn²⁺ ions are absorbed by the negatively charged polysaccharide layer in the cell wall of bacteria and disrupt the balance of the membrane and cause bacterial death.^[36] Xu *et al.*^[22] showed the antibacterial effect of Zn on *S. mutans* attached to the surface.^[34] Petrini *et al.*^[24] showed that ZnO on titanium surface significantly reduced five streptococcus species.^[31] Lui *et al.*^[28] showed that the release of Zn from TiO₂ nanotubes can have continuous antibacterial properties against *S. mutans* and *P. gingivalis.*^[31]

Applications of antibiotics such as amoxicillin, tetracycline, and doxycycline have also been suggested to add antibacterial properties to the surface of implants.^[32] Doxycycline has also shown the characteristics of improving bone growth and treatment of periodontal disease as well as peri-implantitis.^[31,32]

Ferreira *et al.* showed doxycycline coated on nano-tubes implant surface reduced the growth of *P. gingivalis* during 1 month.^[29]

In the third group of the present study, ZnO and doxycycline were added to titanium nano-tubes. The antimicrobial results are compatible with the above-mentioned studies. The inhibitory zones were formed from the first 5-day to fifth 5-day periods. This means that nano-tubes containing doxycycline and zinc oxide, at least until the 1st month, can inhibit the growth of bacteria on the surface of mini-implants and adjacent tissues. The clinical efficacy of this method and the duration of its effective time should be evaluated in future studies.

P. gingivalis, used in this study, is in the red complex and the most potent species in causing periodontal disease,

destroying periodontal support, and inflammation around the implant. It increases in fixed orthodontic treatments and causes significant changes in gingival microbial flora.^[14] The type of microbial growth environments, temperature, incubation period, and the required size are expressed by clinical standards (CLSI). Based on this protocol and the mean of inhibitory zones mentioned in it for different bacteria and antibiotics (approximately less than 30 mm), the inhibitory zone observed in the first 5-day period (48 mm), can be suggested that doxycycline and ZnO have a synergistic effect.

CONCLUSION

1. TiO₂ nano-tubes do not have an anti-microbial effect on adjacent tissues alone and the presence of diffusible agents together with them is necessary.

The formation of nanotubes on the dental mini implant surface, although having antimicrobial properties for the surface, is ineffective for microbes that are far from the surface of the mini implant and cause inflammation and infection in adjacent tissues.

2. The combined application of ZnO and doxycycline as diffusible agents into TiO_2 nano-tubes has an impressive antimicrobial effect. Therefore, it can be suggested as a way to improve the surface of orthodontic mini-implants and promote their success rate.

Declaration of patient consent

Patient consent not required as there are no patients in this study.

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Conflicts of interest

There are no conflicts of interest.

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